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Key indicators

Single-crystal X-ray study
T = 296 K
Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
R factor = 0.042
wR factor = 0.115
Data-to-parameter ratio = 12.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.3-[(2-Hydroxy-5-nitrophenyl)amino]-2-benzofuran-1(3*H*)-one monohydrateThe crystal structure of the title compound, $\text{C}_{14}\text{H}_{10}\text{N}_2\text{O}_5 \cdot \text{H}_2\text{O}$, is stabilized by inversion-related $\text{O}-\text{H} \cdots \text{O}$ and $\text{C}-\text{H} \cdots \text{O}$ intermolecular hydrogen bonds and also by $\pi-\pi$ interactions. The dihedral angle between the phthalide group and the benzene ring is $51.45(8)^\circ$.Received 16 May 2006
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Part XI.

Comment

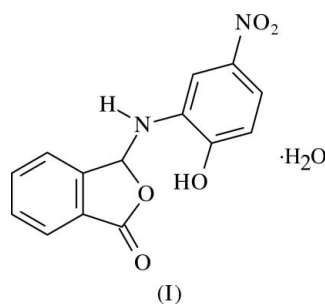
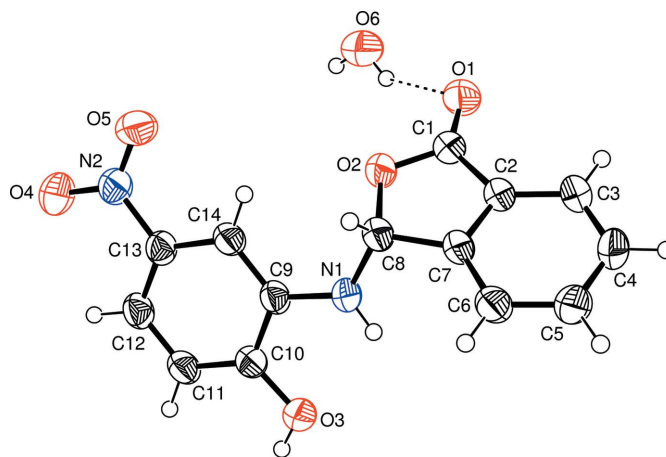
The present work is part of a structural study of compounds of 3-substituted phthalides and we report here the structure of 3-(2-hydroxy-5-nitrophenyl)aminoisobenzofuran-1(3*H*)-one, (I) (Fig. 1 and Table 1).The phthalide group (C1–C8/O2) is planar, the largest deviation from the mean plane being $0.037(1) \text{ \AA}$ for atom O2. The dihedral angle between the mean planes of the phthalide group and the benzene ring is $51.45(8)^\circ$; that between the nitro group and the benzene ring is $2.52(16)^\circ$.

Figure 1

The asymmetric unit of (I), showing the atomic numbering scheme, with displacement ellipsoids drawn at the 50% probability level and a hydrogen bond shown as a dashed line.

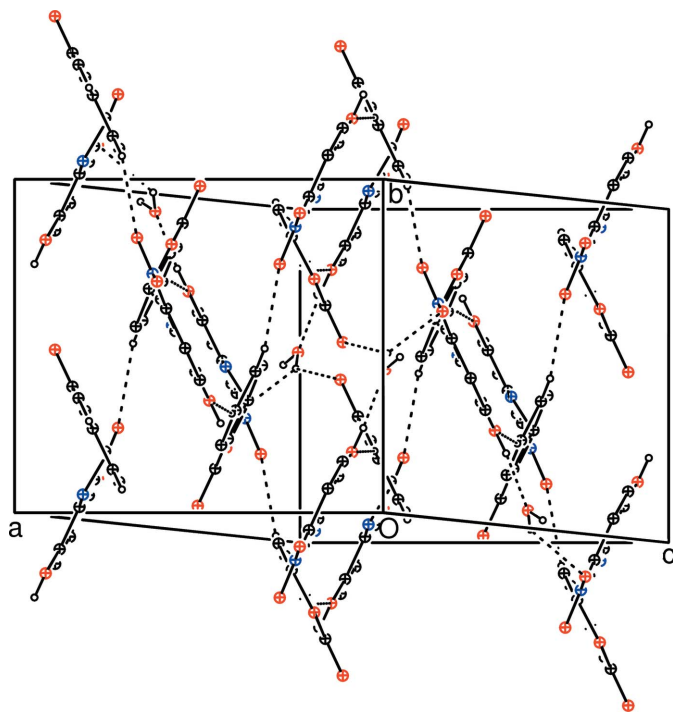


Figure 2

A packing diagram for (I), with hydrogen-bond interactions drawn as dashed lines. H atoms not involved in hydrogen bonding have been omitted.

The crystal packing is stabilized by inversion-related O—H···O and C—H···O intermolecular hydrogen bonds (Fig. 2 and Table 2) and π – π interactions [$Cg1 \cdots Cg1^{vi} = 3.563$ (1) Å; symmetry code: (vi) $2 - x, 1 - y, 1 - z$; perpendicular distance = 3.400 (12) Å; $Cg1$ is the centroid of the C9–C14 ring].

Experimental

The title compound was prepared as described by Odabaşoğlu & Büyükgüngör (2006), using phthalaldehydic acid and 2-hydroxy-4-nitroaniline as starting materials (yield 83%, m.p. 526–527 K). Crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol (95%) solution at room temperature.

Crystal data

$C_{14}H_{10}N_2O_5 \cdot H_2O$
 $M_r = 304.26$
 Monoclinic, $P2_1/c$
 $a = 10.8202$ (7) Å
 $b = 9.8139$ (4) Å
 $c = 15.1475$ (10) Å
 $\beta = 123.588$ (4)°
 $V = 1339.93$ (14) Å³

$Z = 4$
 $D_x = 1.508$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.12$ mm⁻¹
 $T = 296$ K
 Prism, light brown
 $0.45 \times 0.34 \times 0.14$ mm

Data collection

Stoe IPDS-II diffractometer
 ω scans
 Absorption correction: integration
 ($X-RED32$; Stoe & Cie, 2002)
 $T_{min} = 0.952$, $T_{max} = 0.984$

18837 measured reflections
 2628 independent reflections
 2077 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.090$
 $\theta_{max} = 26.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.1115$
 $S = 1.02$
 2628 reflections
 207 parameters
 H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0646P)^2 + 0.1854P]$$

where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{max} < 0.001$
 $\Delta\rho_{max} = 0.24$ e Å⁻³
 $\Delta\rho_{min} = -0.22$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

C1–O1	1.207 (2)	C7–C8	1.501 (2)
C1–O2	1.3502 (19)	C9–N1	1.3949 (19)
C2–C7	1.377 (2)	C13–N2	1.450 (2)
O1–C1–O2	121.19 (14)	N1–C8–O2	112.19 (13)
O1–C1–C2	130.00 (15)		
C14–C13–N2–O4	179.99 (15)		

Table 2

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O3–H3A···O6 ⁱ	0.82	1.84	2.643 (2)	166
O6–H6A···O1	0.790 (19)	2.34 (4)	2.936 (3)	133 (4)
O6–H6A···O4 ⁱⁱ	0.790 (19)	2.36 (4)	3.011 (2)	141 (5)
O6–H6B···O6 ⁱⁱⁱ	0.794 (19)	2.34 (4)	2.937 (4)	133 (5)
C4–H4···O3 ^{iv}	0.93	2.60	3.500 (2)	164
C5–H5···O5 ^v	0.93	2.54	3.367 (2)	148

Symmetry codes: (i) $x, y + 1, z$; (ii) $x, -y + \frac{1}{2}, z - \frac{1}{2}$; (iii) $-x + 2, -y, -z + 1$; (iv) $-x + 1, -y + 1, -z$; (v) $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$.

The water H atoms were refined with distance restraints O–H = 0.83 (2) Å and H6A···H6B = 1.20 (2) Å. All other H atoms were refined using the riding model approximation, with C–H = 0.93 Å for aromatic and 0.98 Å for methine, N–H = 0.86 Å, and O–H = 0.82 Å, with $U_{iso}(H) = 1.2U_{eq}(C,N)$ and $1.5U_{eq}(O)$.

Data collection: $X-AREA$ (Stoe & Cie, 2002); cell refinement: $X-AREA$; data reduction: $X-RED32$ (Stoe & Cie, 2002); program(s) used to solve structure: $SHELXS97$ (Sheldrick, 1997); program(s) used to refine structure: $SHELXL97$ (Sheldrick, 1997); molecular graphics: $ORTEP-3$ for Windows (Farrugia, 1997); software used to prepare material for publication: $WinGX$ (Farrugia, 1999).

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